

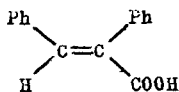
REARRANGEMENT DURING REDUCTION OF CIS- AND TRANS-2-PHENYLCINNAMIC  
ACIDS WITH LITHIUM ALUMINIUM HYDRIDE

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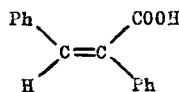
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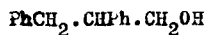
REDUCTION of cis- and trans-2-phenylcinnamic acids, (I) and (II) respectively,<sup>1</sup> with lithium aluminium hydride was expected to give either the saturated alcohol,<sup>2</sup> 2,3-diphenylpropanol (III) or the unsaturated alcohols,<sup>3</sup> cis-2,3-diphenylallyl alcohol (IV) from (I)



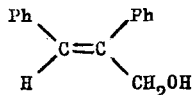
(I)



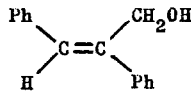
(II)



(III)



(IV)



(V)

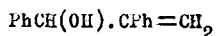
<sup>1</sup> L.F. Fieser, Experiments in Organic Chemistry, 3rd Edition, D.C. Heath & Co., Boston, U.S.A. p. 182 (1955).

<sup>2</sup> Cf. R.F. Nystrom and W.G. Brown, J. Amer. Chem. Soc. **69**, 2548 (1947).

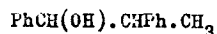
<sup>3</sup> Cf. Li Liang and W.H. Elliott, J. Amer. Chem. Soc. **74**, 4089 (1952).

or trans-2,3-diphenylallyl alcohol (V) from (II) or a mixture<sup>4</sup> of (III) and (IV) or (V).

Treatment of either cis-2-phenylcinnamic acid (I), m.p. 174° or trans-2-phenylcinnamic acid (II), m.p. 137° with excess of lithium aluminium hydride in refluxing ether (4 hours) gave needles (90% from light petroleum, b.p. 40-60°), m.p. 66-67° (Found: C, 85.6; H, 6.7. C<sub>15</sub>H<sub>14</sub>O requires C, 85.7; H, 6.7%). This product has been shown by the following evidence to be 1,2-diphenylallyl alcohol (VI). In ethanol



(VI)



(VII)

the compound has  $\lambda_{\text{max}}$  257m $\mu$  (log  $\epsilon$  4.09) typical of a styrene<sup>5</sup> but not of a stilbene system.<sup>6</sup> Ozonolysis gave benzil in 94% yield.

Hydrogenation over Adams catalyst at room temperature and atmospheric pressure gave 1,2-diphenylpropanol (VII), characterised as its phenylurethane, m.p. 115-116°.<sup>7</sup>

1,2-Diphenylallyl alcohol (VI) was recovered after treatment with acetic anhydride for several hours followed by hydrolysis with ethanolic potassium hydroxide, conditions which normally cause anionotropic rearrangement of 1-phenylallyl alcohols.<sup>8</sup> This result indicates that (VI) is thermodynamically more stable than either (IV) or (V), possibly a result of steric inhibition of resonance in the uniplanar conformations of (IV) and (V).<sup>9</sup>

<sup>4</sup> Cf. F. Weygand and D. Tietjen, Chem. Ber. **84**, 625 (1951).

<sup>5</sup> R.A. Morton and A.L. Stubbs, J. Chem. Soc. 1347 (1940).

<sup>6</sup> E.A. Braude, J. Chem. Soc. 1902 (1949).

<sup>7</sup> F. Kayser, Ann. Chim. (France) **6**, 145 (1936).

<sup>8</sup> H. Burton and C.K. Ingold, J. Chem. Soc. 904 (1928).

<sup>9</sup> Cf. E.A. Braude and P.H. Gore, J. Chem. Soc. 41 (1959).